

# NU-LHT-1M

*NASA/USGS - LUNAR HIGHLANDS TYPE – v.1  
MEDIUM-GRAINED (<1mm)*

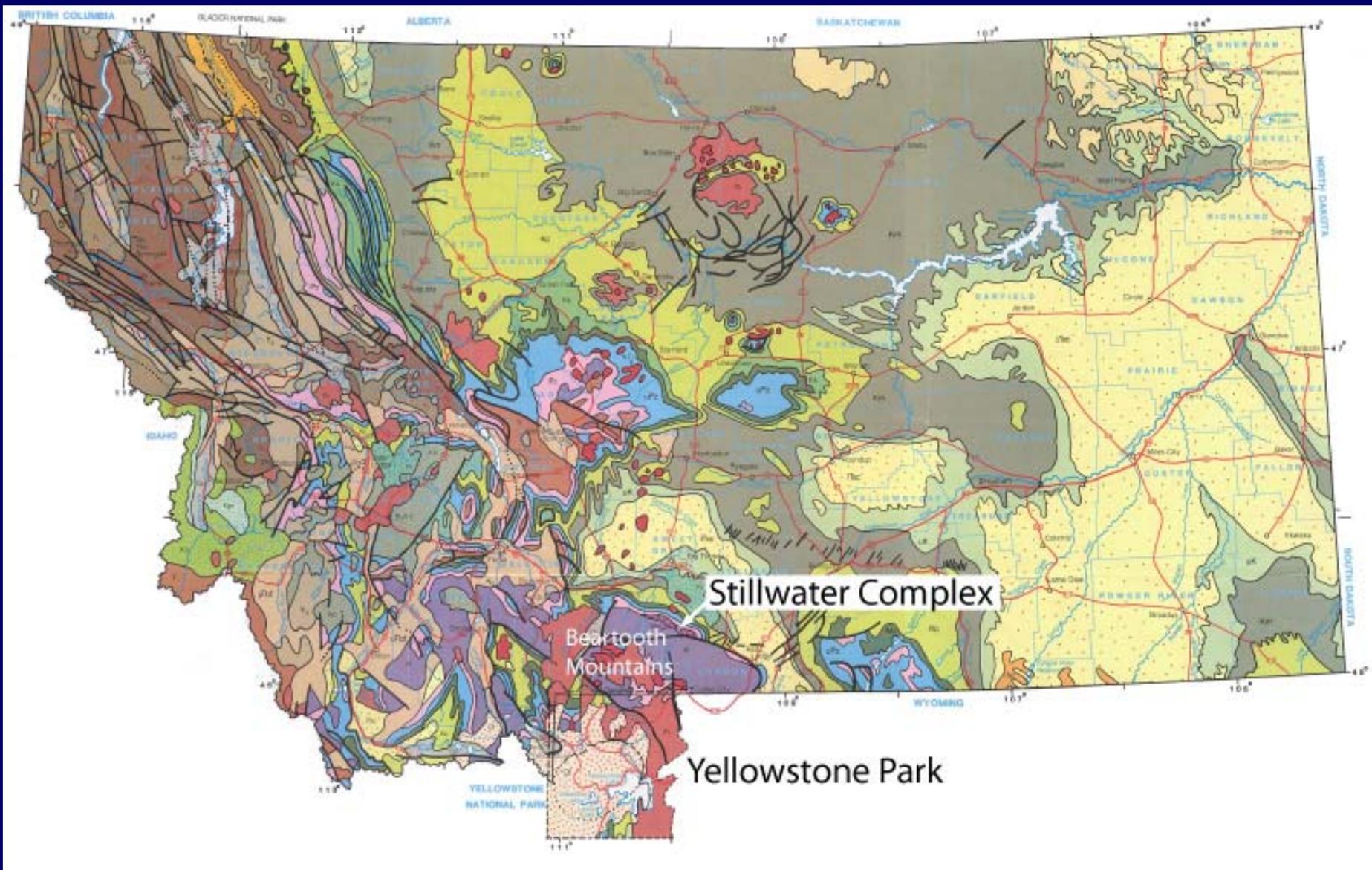
## PILOT

# HIGHLANDS SOIL SIMULANT

Douglas Stoesser & Steve Wilson  
U.S. Geological Survey, Denver, CO

# WHY & WHAT

- Why LHT simulants? If we put a base at the poles, the base will be on a highlands type regolith and therefore highlands regolith simulants are needed.
- We used an average Apollo 16 regolith target composition.
- The bulk of our material came from the Stillwater Complex in Montana which has long been proposed as a source of materials for highland simulants.



Stillwater Complex

Beartooth Mountains

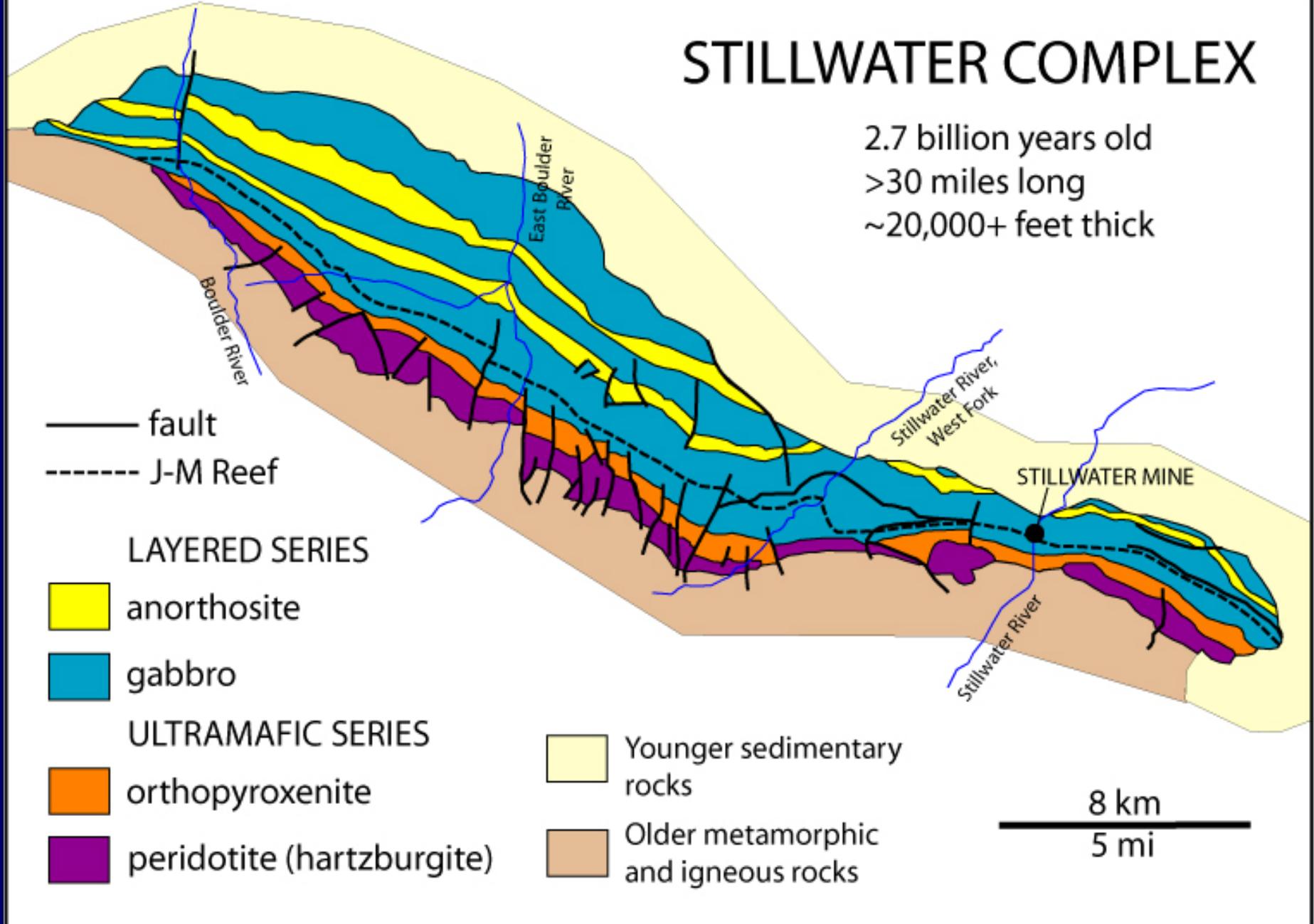
Yellowstone Park

# LHT-SOURCE MATERIALS

- The Stillwater Mining Company is actively supporting the simulant effort. The SMC is mining platinum and palladium from the Complex in the basal norite zone (from the "J-M Reef", 6-9 ft wide & 28 miles long).
- Both the highlands and the Stillwater are composed of the same major minerals: plagioclase (Na-Ca-Al-Silicate), pyroxenes and olivine (Fe-Mg silicates).
- The lunar highlands regolith has the overall composition of a anorthositic gabbro-norite (~75-83% calcic-plagioclase, ave. ~ An94).
- The Stillwater Complex is a mafic layered intrusion consisting of hartzburgite, orthopyroxenite, norite, gabbro-norite, gabbro, and anorthosite.
- LHT-1M is completely composed of rock materials from the Stillwater Complex, except for ilmenite obtained courtesy of Iluka Resources.
- LHT-1M glasses were made from Stillwater Mill waste.

# STILLWATER COMPLEX

2.7 billion years old  
>30 miles long  
~20,000+ feet thick



— fault  
- - - J-M Reef

## LAYERED SERIES

 anorthosite  
 gabbro

## ULTRAMAFIC SERIES

 orthopyroxenite  
 peridotite (hartzburgite)

 Younger sedimentary rocks  
 Older metamorphic and igneous rocks

8 km  
5 mi

# Bear Tooth Mountains

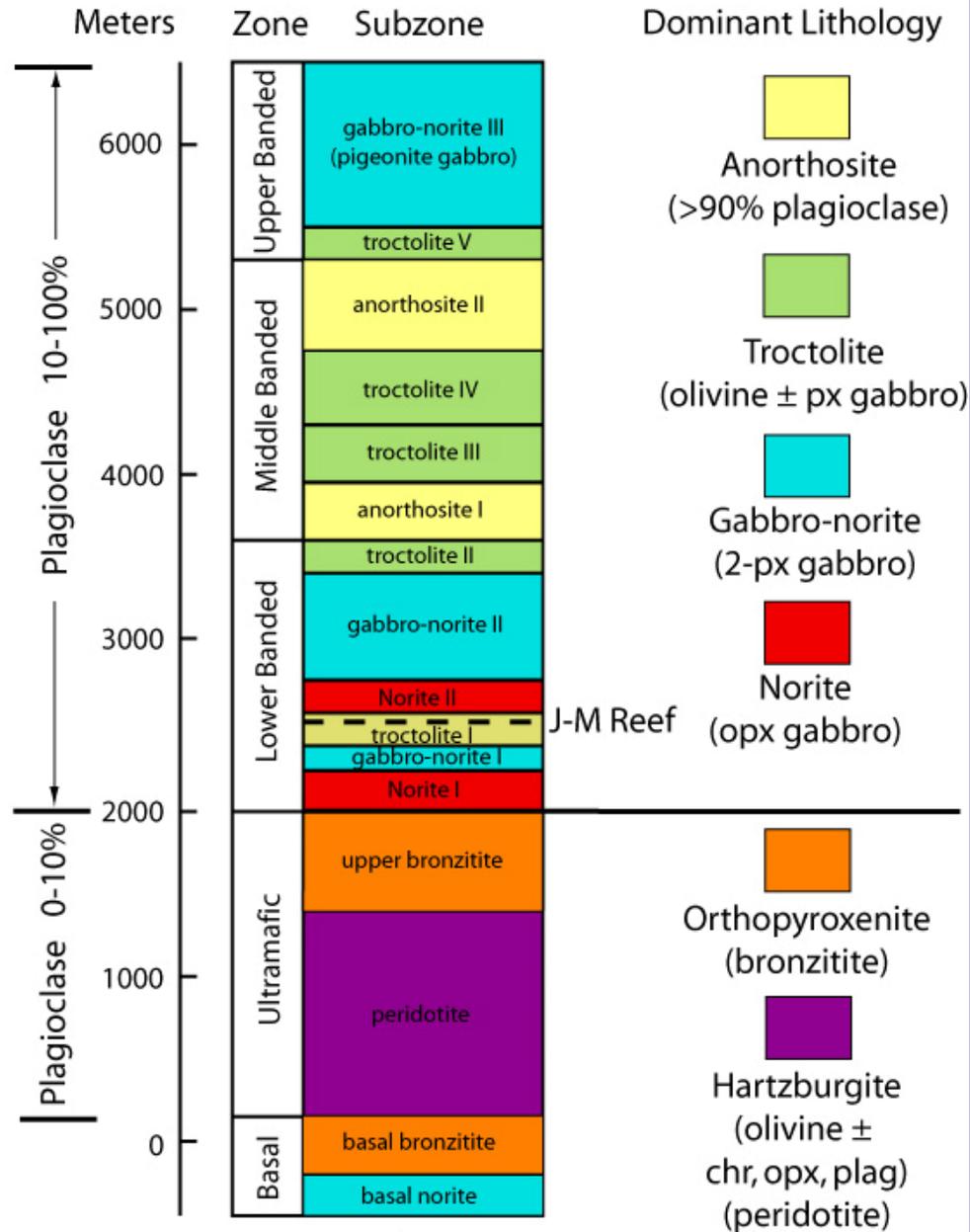
Complex

Stillwater Mine

Stillwater



# STILLWATER COMPLEX



After McCallum et al. (1980)

# Mountain View Mine

Hartzburgite &  
Orthopyroxenite

Chromite

Anorthosite

Mill Portal

Waste Dump  
(Norite & Anorthosite)

Stillwater River

## STILLWATER MINE

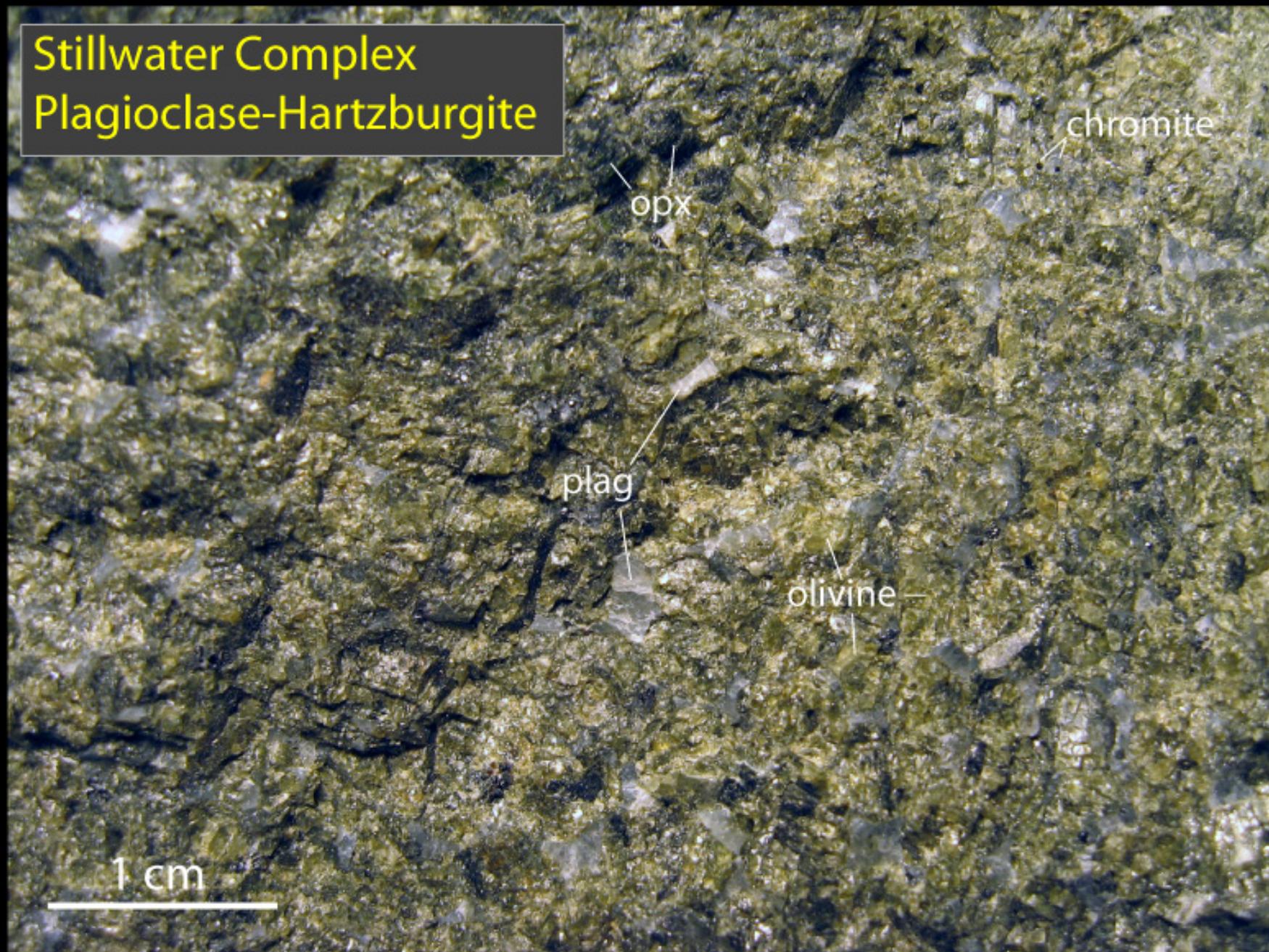




1 cm

Chromite-Hartzburgite  
(olivine cumulate)

Stillwater Complex  
Plagioclase-Hartzburgite





cpx

opx

pl

1 cm

Stillwater  
Plagioclase-Orthopyroxenite

# NORITE



PLAGIOCLASE (WHITE)  
ORTHOPYROXENE (BROWN)

NORITE FEED STOCK





## STILLWATER MILL "SAND" (WASTE)

Composition of plagioclase-rich  
gabbro-norite (69% normative  
plagioclase, AN82)

# ANORTHOSITE

poikilitic  
clinopyroxene



1 cm



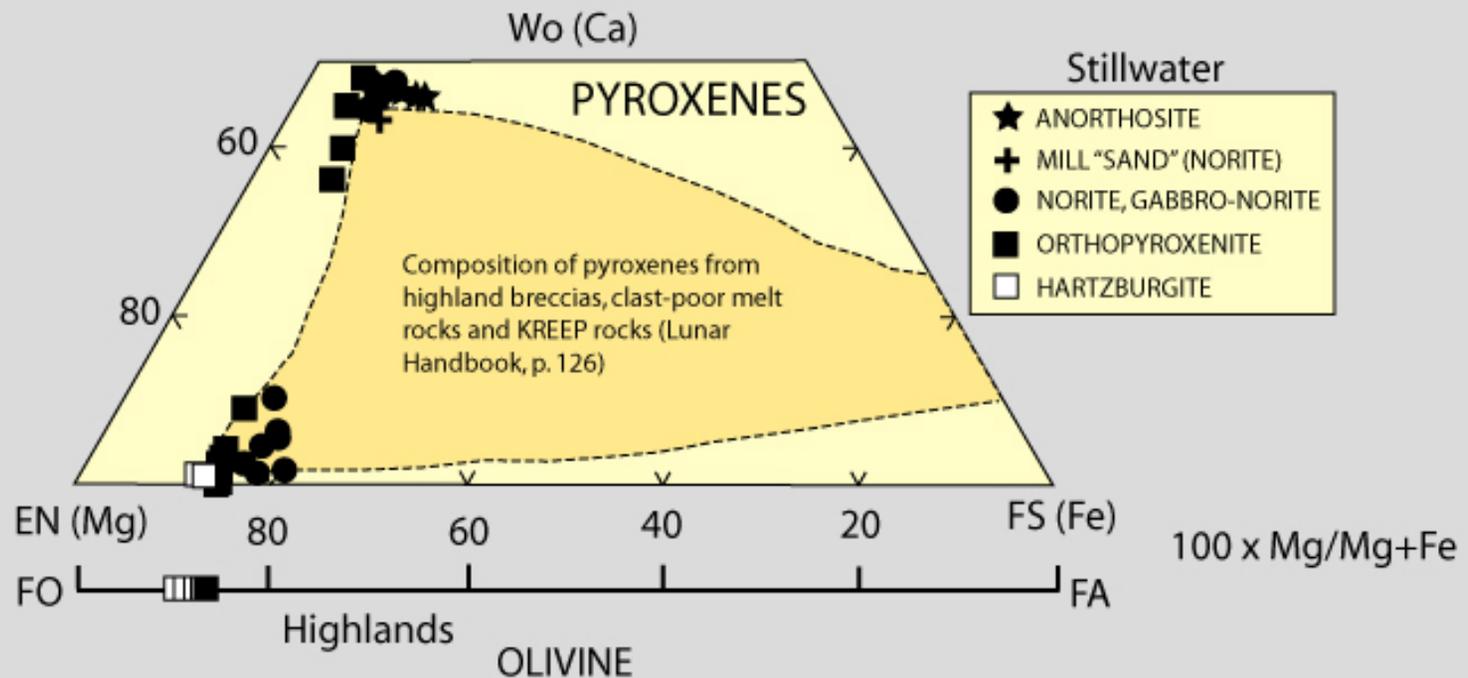
# LHT-1M Component Mineral Compositions

## PLAGIOCLASE (ave. AN content & range)

Anorthosite:	79	(62-85)
Mill Sand	83	(60-96)
Norite:	84	(78-92)
Orthopyroxene	75	(73-78)
Hartzburgite	66	(64-68)

$$AN = 100 \times Ca / (Ca + Na)$$

Average highlands ~AN 94



# Simulant requirements

- Proper Mineralogy
  - Appropriate chemical composition
  - Absence of alteration minerals
- Optimal particle size distribution
- Appropriate shape distribution
- Major glass content
  - High quality (HQ) glass
  - Agglutinate (LQ) glass

# Development approach

- Target average Apollo 16 lunar soil composition  
NASA conference pub 2255 (1982)
- Chemical analysis of starting constituents
- USGS blending program optimizes mixture
- Mixture is modified to include best mineralogical composition

# Chemical composition starting materials

<u>Sample ID</u>	<u>SiO<sub>2</sub></u>	<u>Al<sub>2</sub>O<sub>3</sub></u>	<u>Fe<sub>T</sub>O<sub>3</sub></u>	<u>MgO</u>	<u>CaO</u>	<u>Na<sub>2</sub>O</u>
Norite	49.4	23.72	4.07	8.17	13.6	0.84
Anorth.	48.6	30.9	1.54	1.47	15.9	1.7
Harz.	43.7	2.1	<b>13.1</b>	<b>38.3</b>	1.87	<0.2
<b>Sand</b>	47.7	22.1	6.2	10.5	12.5	0.9
Apollo 16	44.9	25.0	9.3	5.8	15.7	0.47

# Components used

<u>Material</u>	<u>Amount</u>	<u>% of total</u>
Norite	38 lbs	30.0
Anorthosite	55	43.4
Hartzburgite	7.7	6.1
Ilmenite	0.92	0.73
Glass	25	<b>19.7</b>
HQ glass	5	(20)
LQ glass	20	(80)

# Simulant preparation

- Starting material crushed <4 mm  
Norite, Anorthosite, Hartzburgite,
- Grinding time experiments (ball mill)  
Consistent size reduction
- Develop appropriate mixture of starting material
- Grind for specific time period (15, 45 min.)
- Blend as one batch
- Split using spinning riffler (200g aliquots)

# LHT-1M composition

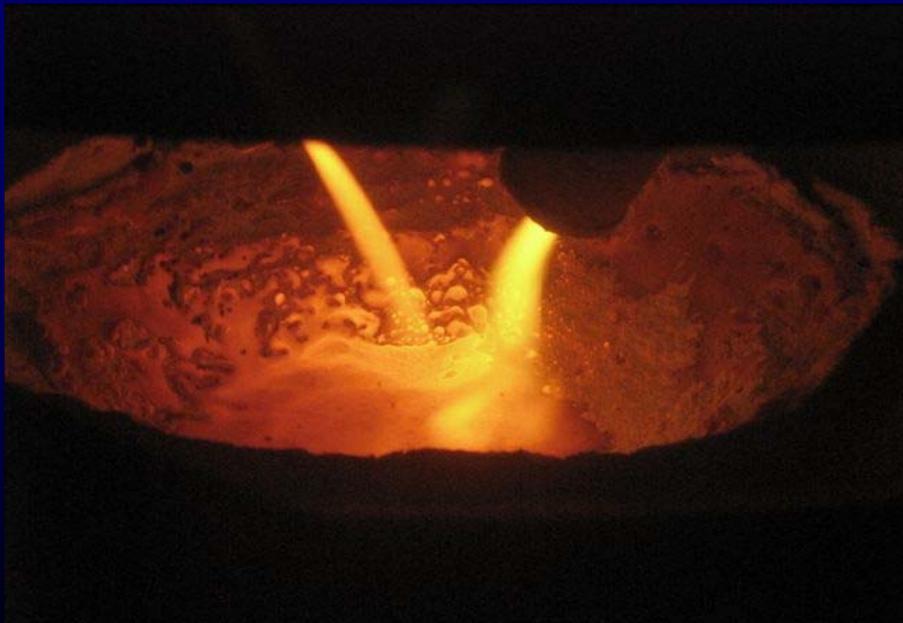
<u>Oxide</u>	<u>Apollo 16</u>	<u>LHT-1M</u>
SiO <sub>2</sub>	44.9	47.6
Al <sub>2</sub> O <sub>3</sub>	25.0	24.4
FeO	8.4	4.3
MgO	5.8	8.5
CaO	15.7	13.1
Na <sub>2</sub> O	0.47	1.4

# Glass production

- Need to develop technology for glass production
  - HQ glass
  - Agglutinate (LQ) glass
- High throughput 100's kg per day
- Excellent temperature control
- Able to handle geologic source material (powder)
- High temperature plasma

# Plasma Technology

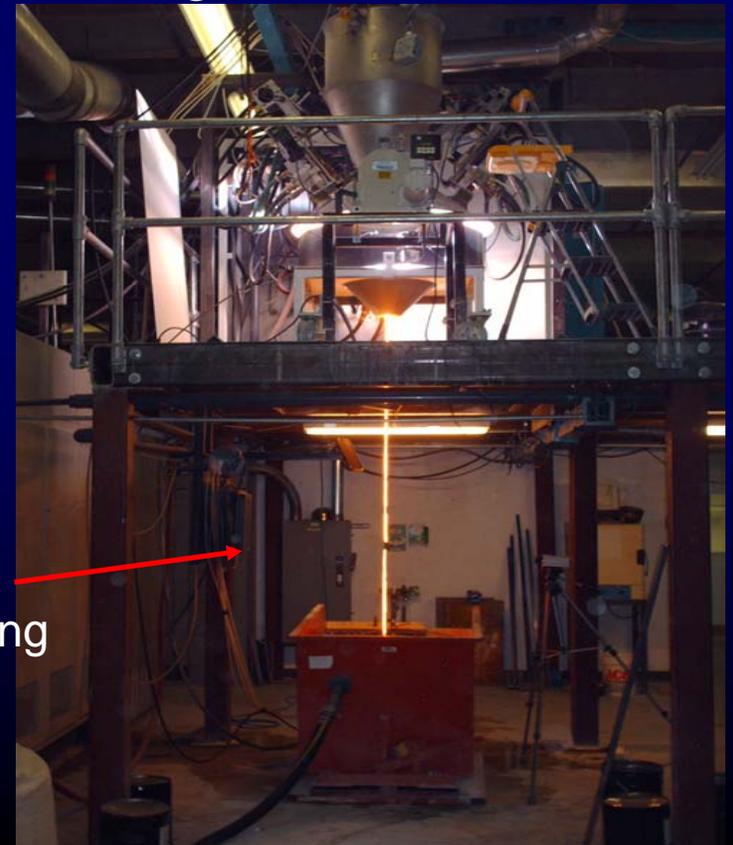
- Remotely-coupled transferred arcs
  - Non-conductive material rapidly heated / cooled
  - Extreme thermal gradient
  - Power density -  $140 \text{ MW/m}^3$
- Glass exit temperatures  $1,300\text{--}1,925 \text{ }^\circ\text{C}$ . Up to  $150 \text{ kg/hr}$
- Plasma temperatures  $13,000\text{ }^\circ\text{K}$  to  $21,000\text{ }^\circ\text{K}$ 
  - Typically run at 600 to 1,200 amps



LHT glass  
manufacturing



Zybek Advanced Products | Z  
Product Engineering, Research & Development | P

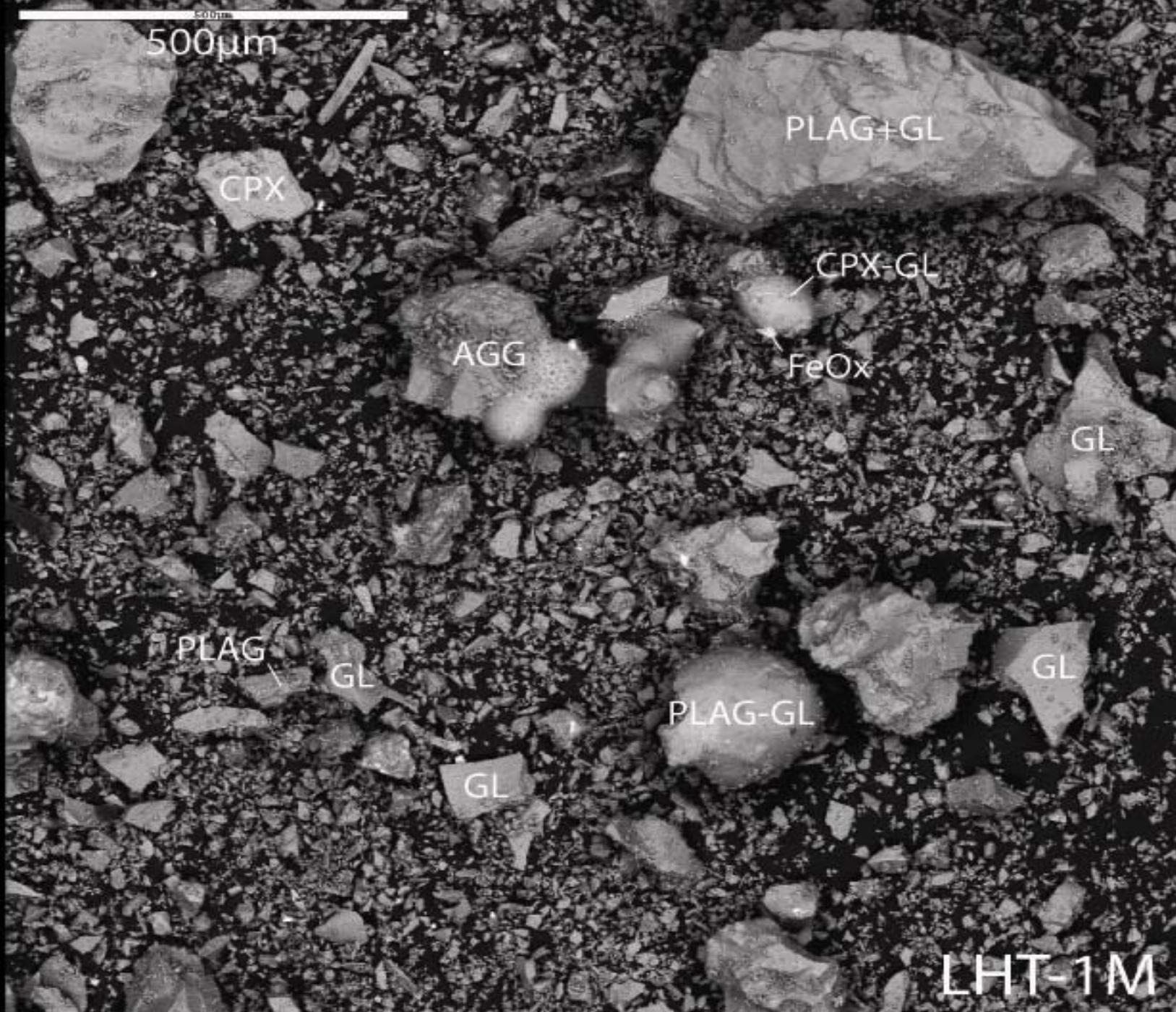


# Agglutinate production

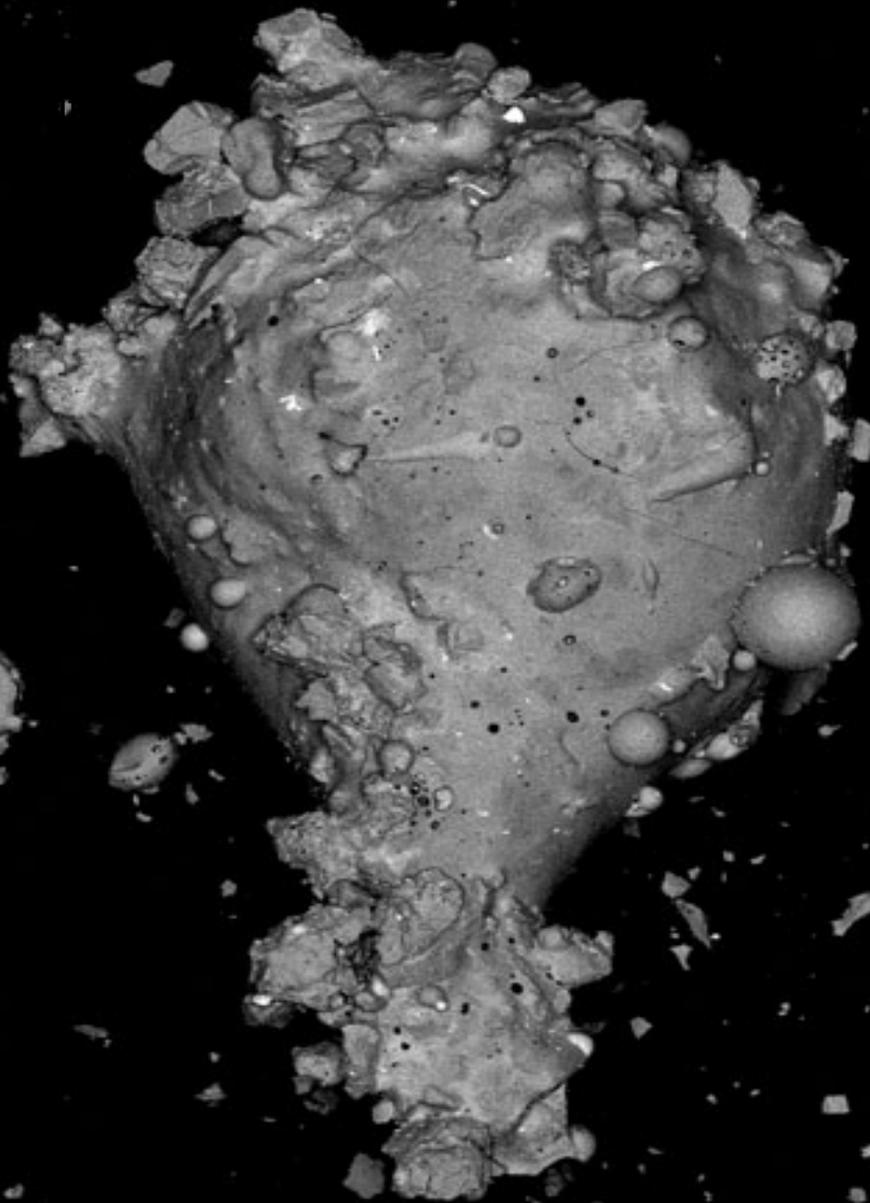
- Partially melted material
- Irregular internal structure, vesicles
- Very friable
- Production rates at 50+ kg (110+ lbs) per day



# Movie clip



to



15kU

X37 500µm

USGS LHT-1

200μm

PLAG

PLAG-GL

ROCK

NORITE-GL

PLAG-GL

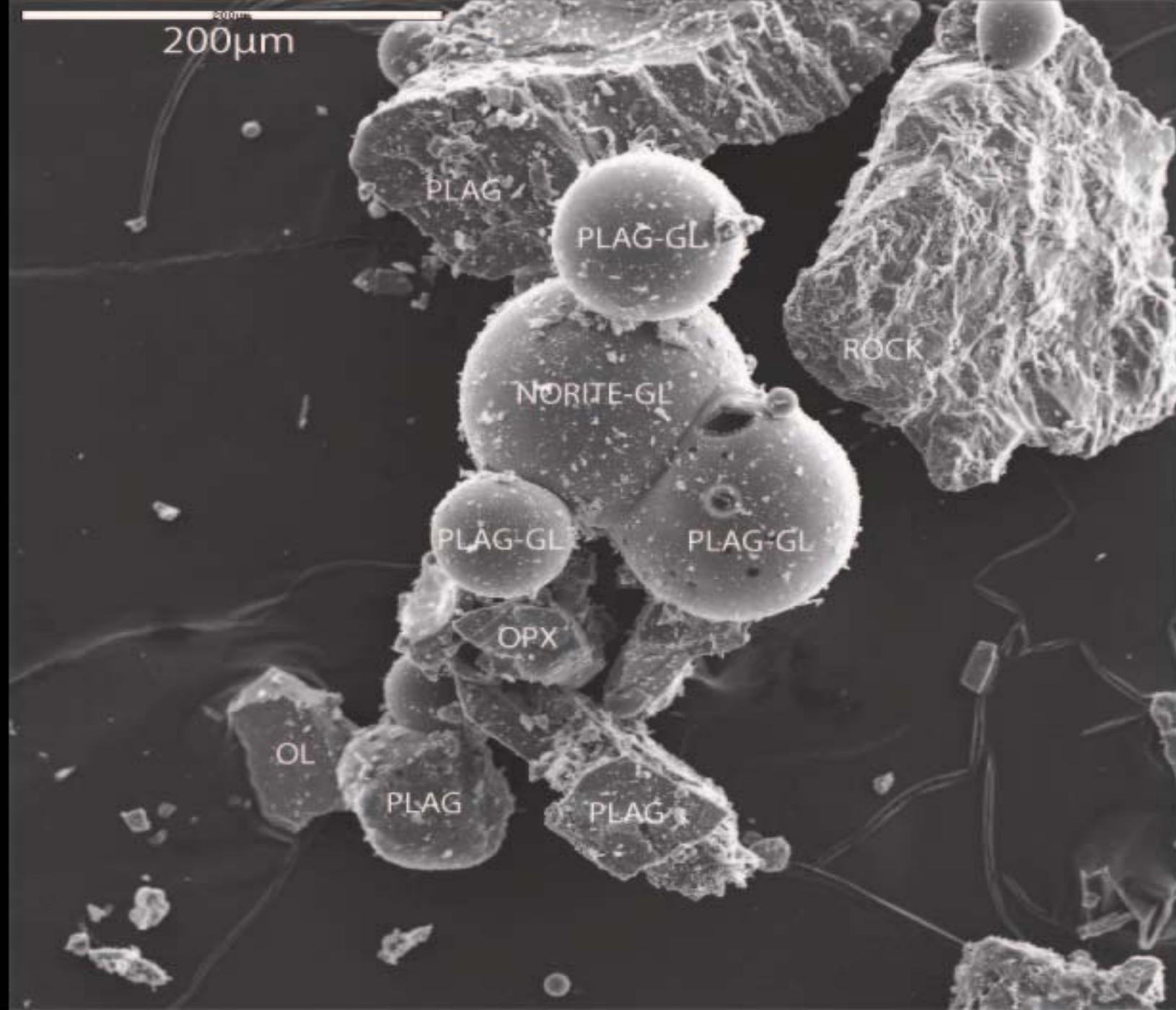
PLAG-GL

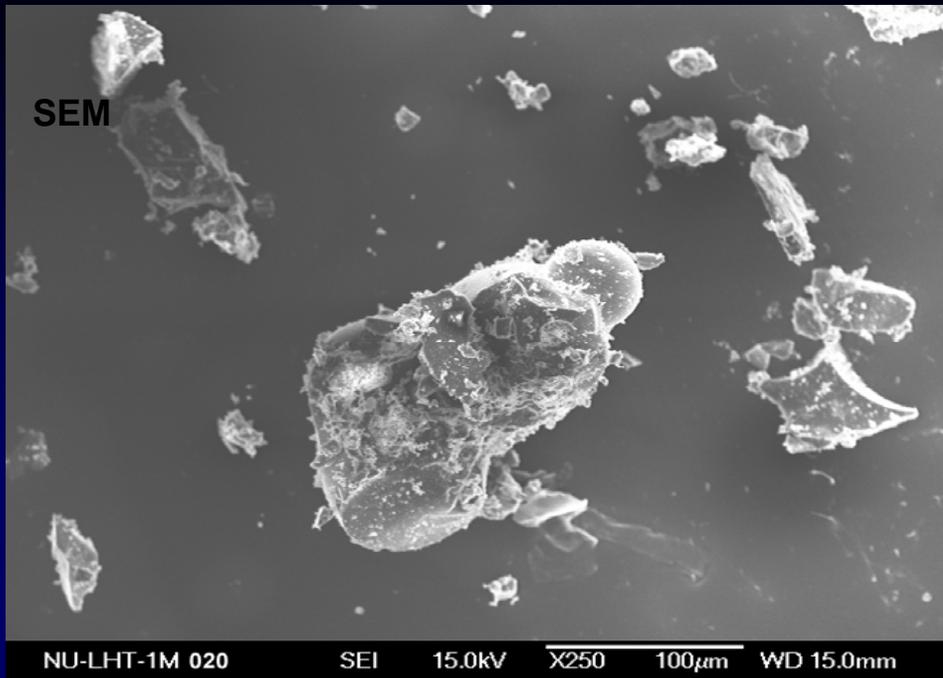
OPX

OL

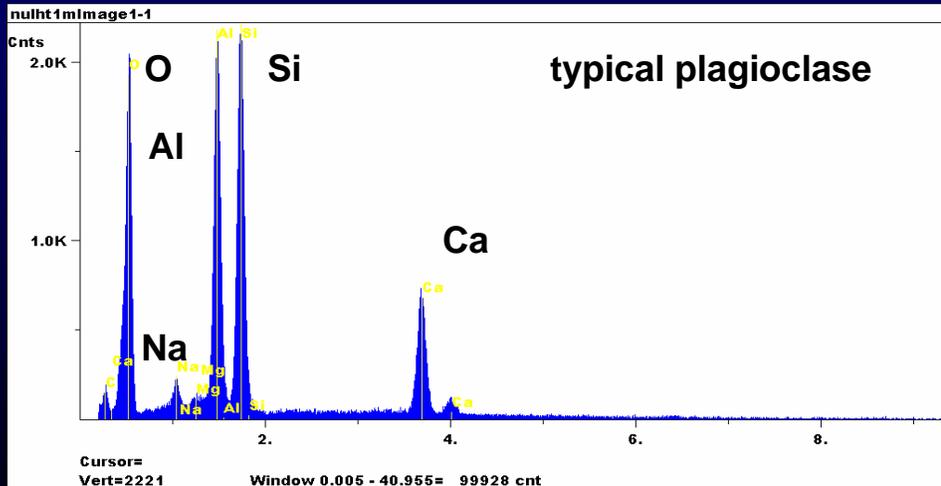
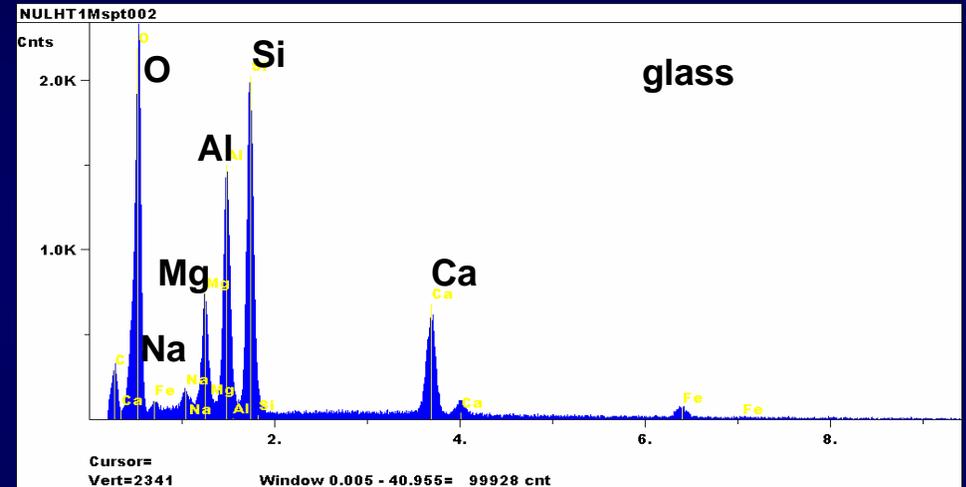
PLAG

PLAG





Glass spherules are rare to isolated. This one is firmly attached to larger vesicular glass. Glass has Mg-rich highlands composition regardless of shape, and is homogeneous. No inclusions have been identified in glass. Nothing found in sample that looks like agglutinate. Found isolated ilmenite(?), a Zr-rich grain, and a grain consisting of Si, Al, C, and O.



Glass constituents (in order of decreasing peak height, which qualitatively correlates to abundance): O, Si, Al, Mg, Ca, Na, Fe.

Composition is excellent approximation to those of many lunar highland impact glasses.



# System overview

Carl Zeiss custom SEM



QEMSCAN® uses advanced e-beam technology from Carl Zeiss and combines this with high resolution BSE and SE imaging, and state-of-the-art Energy Dispersive Spectrometers. It integrates these using iDiscover software to provide a solution capable of identifying most rock-forming minerals in just milliseconds

Bruker AXS X-ray detectors



Intellecion iDiscover™ analytic software suite

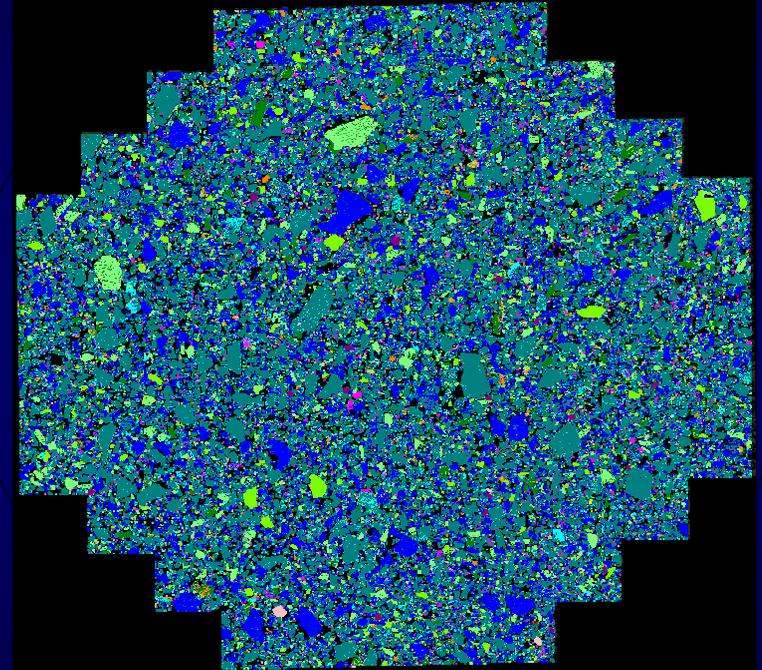
Intellecion QEM\*SEM® integrated control hardware



# QEMSCAN<sup>®</sup> Results



Digital photograph of a 30mm diameter polished block

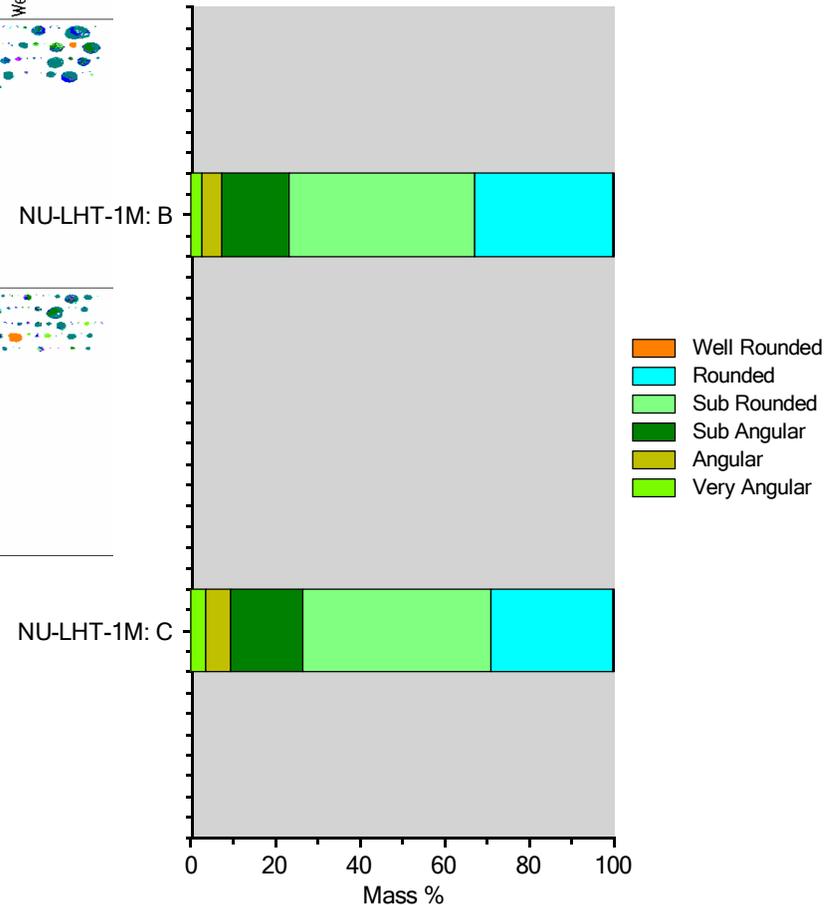
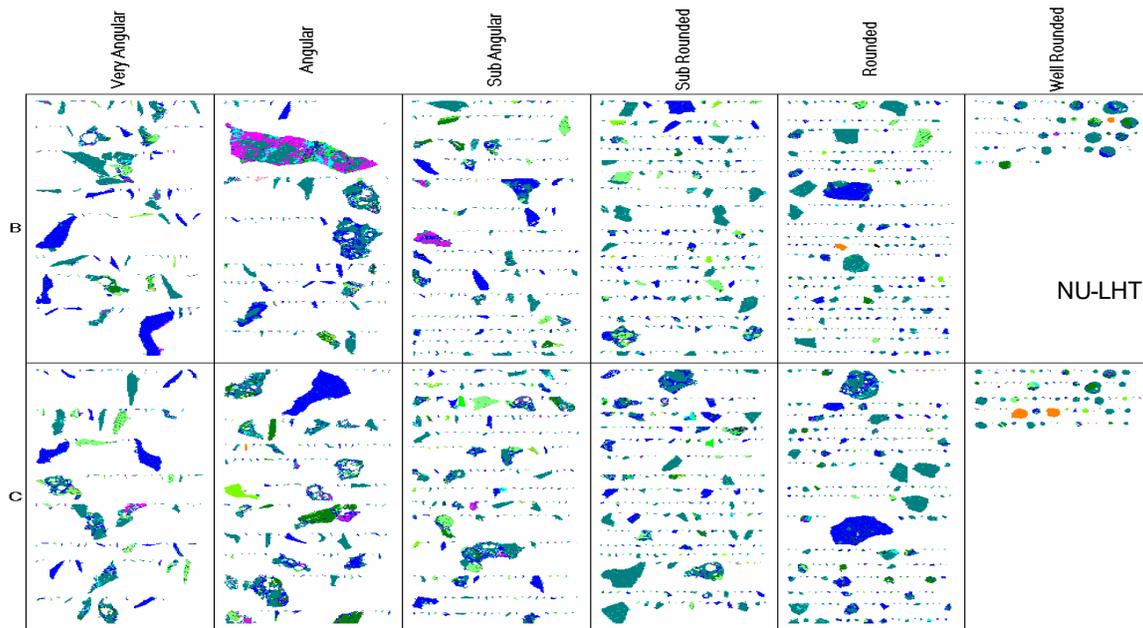


QEMSCAN<sup>®</sup> false-coloured, digital particle mineral map montage of a polished block

# Modal Composition and Repeatability

	NU-LHT-1M: C	NU-LHT-1M: B
CaMgFe Glass	26.0	26.8
MgFeAl Silicate	0.5	0.5
Ca Al Silicate Glass	1.7	2.2
Mg Silicates	1.3	1.4
Olivine	3.5	3.6
Orthopyroxene	7.6	6.8
Clinopyroxene	8.3	8.2
K Feldspar	0.1	0.1
Plagioclase (An60-An100)	45.6	44.6
Plagioclase (Ab-An60)	4.1	4.1
Quartz	0.2	0.2
Ilmenite	0.6	0.4
Chromite	0.1	0.1
Magnetite	0.1	0.3
Iron	0.0	0.5
Others	0.3	0.4
<b>Total</b>	<b>100</b>	<b>100</b>

# Simulant Particle Shape - Quantified



	NU-LHT-1M: C	NU-LHT-1M: B
Very Angular	3.5	2.5
Angular	5.8	4.8
Sub Angular	17.2	15.9
Sub Rounded	44.3	44.0
Rounded	29.1	32.6
Well Rounded	0.2	0.2
<b>Total</b>	<b>100</b>	<b>100</b>

# Production plans

- Create a prototype material NU-LHT-2M
  - Improvements from LHT-1M effort
  - Addition of minor minerals - fidelity
  - Nanophase iron in agglutinate?
  - More spheroids
- 900 – 1000 pounds
- Expected completion date 1-1-08



U.S. Geological Survey  
National Aeronautical and Space Administration  
*Certificate of Analysis*

**Lunar Highlands Type simulant (Pilot)**  
**NU-LHT-1M**

Material used in the preparation of this pilot simulant material was collected under the direction of the USGS in 2007 from the Stillwater Mining Company, located near Nye, Montana. A variety of samples from the mining operations and locations adjacent to the mine were obtained which are believed to represent the various geologic rock types found in the highlands region of the moon. The USGS performed a basic separation of the starting material by visual inspection, followed by blending of selected aliquots based on chemical and mineralogical considerations. Glass and agglutinate production using material from the Stillwater mine was performed at Zybeck Technologies Inc. using a high temperature plasma melter. Glass and mafic components were then combined, processed and bottled at the USGS. Information presented below is based on analyses performed at the USGS, Bureau of Reclamation, and Horiba Instruments Inc.

<u>Element</u>	<u>Mean</u>	<u>s.d.</u>	<u>Oxide</u>	<u>Mean</u>	<u>s.d.</u>
Al, %	12.9	1.0	Al <sub>2</sub> O <sub>3</sub> , %	24.4	3.5
Ca	9.38	0.40	CaO	13.1	0.56
Fe (II)	2.55	0.01	FeO	3.30	0.012

# Acknowledgements

- Stillwater Mining company - Dave Ryckman
- JSC collaborators
  - John Lindsay, Sara Nobel, Susan Wentworth
- Horiba instruments, shape/size analysis
- Zybeck Advance Products – Mike Weinstein
- Intellection - Alan Butcher